A HIGHLY EFFICIENT AND STEREOSELECTIVE CYCLOGLYCOSYLATION. SYNTHESIS OF $CYCLO(-4)-(a-Man-(1\rightarrow 4))$ ₅- α -Man- $(1\rightarrow)$, A MANNO ISOMER OF α -CYCLODEXTRIN

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Abstract: Stereocontrolled synthesis of a manno isomer of α -cyclodextrin was achieved for the first time employing PhSeOTf promoted cycloglycosylation of octadeca-O-benzyl- α -Man- $(1\rightarrow 4)$ - $\{\alpha$ - $Man-(1\rightarrow 4)$]_{4 - α -Man-1 \rightarrow SMe.}

Cycloglycosylations leading to the formation of cyclooligosaccharides have been reported only in few cases¹. In the field of cyclooligosaccharides, synthetic studies have mostly been directed to the functional group manipulations² of cyclodextrins (cyclomaltooligoses). Recently we reported a first chemical synthesis³ of α -cyclodextrin (cyclomaltohexaose) 1 via octadeca-Obenzylether 3 that was prepared in 21% yield by SnCl2-AgOTf4 promoted cycloglycosylation of the key intermediate 5. Efficiency of cycloglycosylation depended *on* the ring size and in the case of cyclomaltooctaose (y-cyclodextrin) cycloglycosylation of the corresponding maltooctaosyl fluoride under the same condition could be achieved only in 8.4% yield⁵.

In close connection with these synthetic studies on cyclooligosaccharides, we report here an efficient and practical cycloglycosylation of a mannohexaosyl methylthioglycoside 6 that led to a stereocontrolled synthesis of the unnatural cyclooligosaccharide, cyclo α -(1-+4)-linked

mannohexaose 2, a manno isomer of α cyclodextrin. Thus PhSeOTf promoted⁶ cycloglycosylation of methylthioglycoside 6 proceeded at -20 $^{\circ}$ in (CH₂Cl)₂ to afford a 64% yield of octadeca-O-benzyl cyclomannohexaose 4^7 , which was smoothly hydrogenolized in the presence of 10% Pd-C in MeOH to give desired 2^7 . Surprisingly, use of p -methyl benzoyl group as an O-2 stereocontrolling auxiliary 8 in place of an O-2 benzyl group in compound 6 deteriolated the smooth cycloglycosylation. Attempted cycloglycosylation of compound 7 gave multiple products under various conditions. A reasonable explanation for this unexpected observation has remained to be made.

The two key intermediates 6 and 7 were synthesized according to Scheme 2 by employing a mannotriaosyl donor 8 carring

Scheme 2 $(CA = C)CH₂CO$, MP = p-MeOPh)

p-methyl benzoyl group at O-2 as a stereocontrolling auxiliary and a mannotriaosyl acceptor 9, which in turn were prepared starting from monosaccharide synthons 10 and 11.

A glycosyl acceptor 10^7 was designed to carry p-methoxyphenyl group for the purpose of temporary protection of the anomeric hydroxy function and was prepared in 4 steps in 54% overall yield from pentaacetyl mannose 12 via 137 (1 p-MeOPhOH-CF3SO3H in (CH2Cl)2, 2 NaOMe in MeOH, 3 $(Bu_3Sn)_2O^9$ then BnBr-Bu4NBr¹⁰, 4 p-MeBzCl in pyridine at -20°). Conversion of 10 into a mannosyl donor 11 was performed in a straightforward manner in 3 steps via 14⁷in 50% overall yield (1 (CICH₂CO)₂O-NaHCO₃¹¹ in DMF, 2 (NH₄)₂Ce(NO₃)₆ in 4:1 CH₃CN-H₂O¹², 3 CCl₃CN-DBU in $(CH_2Cl)_2^{13}$). TMSOTf promoted glycosylation¹⁴ of the glycosyl acceptor 10 with the trichloroacetimidate 11 according to Schmidt¹³ in the presence of powdered molecular sieves $4A$ proceeded stereoselectively to give an 87% yield of α -(1->4) linked mannobioside 16⁷, which was treated with (NH_2) ₂CS in EtOH¹⁵ to give selectively dechloroacetylated product 17⁷ in 98% yield. The mannobiosyl glycosyl acceptor 17 was smoothly converted into mannotriaosyl glycosyl donor 8^7 in 3 steps in 56% overall yield via 187 and 197 (1 11, TMSOTf and MS4A in (CH2Cl)2 at -20°, 2 (NH₄)₂Ce(NO₃)₆ in 4:1 CH₃CN-H₂O, 3 Cl₃CCN-DBU in (CH₂Cl)₂ at -20°). Mannotriaosyl glycosyl donor 97 was obtained from 8 in 2 steps in 78% overall yield via 20^7 (1 Bu3SnSMe¹⁶-BF3·Et₂O in $(CH_2Cl)_2$, 2 (NH₂)₂CS in EtOH).

Having both glycosyl donor 8 and acceptor 9 prepared in a stereocontrolled way, coupling between them was examined in the presence of TMSOTf to afford an 87% yield of mannohexaosyl derivative $.21⁷$ which was then dechloroacetylated to give an 81% yield of the designed key intermediate 7^7 . Available procedures for the activation of thioglycoside 7 by use of thiophilic reagents failed to give any isolable amount of cyclyzation product. Therefore, completely

benzylated derivative 6 was chosen as an alternative key intermediate for cycloglycosylation. Replacement of O-2 p-methyl benzoyl groups of 7 into O-2 benzyl groups was achieved in 4 steps to give 6⁷ in 75% overall yield via 22⁷ and 23⁷, (1 EtOCH=CH₂ and PPTS in (CH₂Cl)₂, 2 NaOMe in MeOH, 3 BnBr and NaH in DMF, 4 Amberlyst 15 resin in 1:l CHCl3-MeOH).

Crucial cycloglycosylation of 6 was achieved successfully as already described. A higher efficiency of the cycloglycosylation for manno 0.Rn compound 6 compared with that for $gluco$ isomer 5 may not be explained only by the difference of anomeric **BnO** learing groups. The *manno* configuration at the Bn O

reducing-end residue of the "intimate ion-pair" 17 intermediate 25 might play important role for this result.

In summary, an efficient synthesis of a *manno* isomer of α -cyclodextrin was excecuted in 3.4% overall yield in 21 steps from readily obtainable pentaacetylmannopyranose 12 by use of PhSeOTf promoted cycloglycosylation of the key thioglycoside 6.

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- 7 Physical data for key compounds are described below. Values of $[\alpha]$ and δ H, C were measured for CHCl3 and CDCl₃ solutions, respectively, at 25°, unless noted otherwise. 2: α]_D +26.5° (c 0.2, H₂O); δ _H(D₂O) 4.926 (d, 2.4 Hz, H-1), 4.009 (t, 3 Hz, H-2), 3.944 (dd, 2.5, 11.3 Hz, H-6), 3.925 (dd, 3.4, 9.0 Hz, H-3), 3.880 (dd, 5.2, 12.2 Hz, H-6), 3.787 (ddd, 9.5, 2.4, 4.9 Hz, H-5), 3.703 (t, 9.0 Hz, H-4). 4: [a]D -2.8° (c 0.3); δ H 5.043 (d, 1.5 Hz, H-1); δ C 100.9 (1 JCH 165 Hz, C-1). 6: [α]D -2.1° (c 0.8); δ H 5.333 (bs, H-1 x 2), 5.290 (bs, H-1 x 4), 2.172 (s, SCH3). 7: $[\alpha]_D$ -52.6° (c 0.9); δ_H 5.572 (bs, H-1), 5.562 (bs, H-1 x 2), 5.546 (bs, H-1 x 2), 5.331 (d, 1.5 Hz, H-1a), 2.220 (s, SCH₃); δ C 99.4, 99.1, 99.0, 98.9, 98.8 (C-1bcdef), 83.9 (¹J_{CH} 164 Hz, C-1a). 8: $[\alpha]_D$ -21.2° (c 0.7); δ_H 6.420 (d, 2.1 Hz, H-1a), 5.571 (d, 1.8 Hz, H-1 x 2), 5.484 (t, 10.0 Hz, H-4c), 3.723 (s, CH₂Cl). 9: [a]_D -25.5° (c 0.4); δ H 5.545 (d, 1.5 Hz, H-1), 5.510 (d, 1.8 Hz, H-1), 5.324 (d, 1.5 Hz, H-1a), 2.208 (s, SCH₃); δ C 99.4 (¹J_{CH} 175 Hz, C-1bc), 83.8 (¹J_{CH} 169 Hz, C-1a). **10**: [α]_D +10.5° (c 0.8); δH 5.747 (dd, 1.8, 3.1 Hz, H-2), 5.565 (d, 1.8 Hz, H-1), 3.758 (s, OCH3). 11: [α]D -20.1° (c 1.9); δH 6.414 (d, 1.2 Hz, H-1); δ C 95.1 (¹J_{CH} 179 Hz, C-1), 40.6 (COCH₂Cl). 13: [a]_D +73.9° (c 1.9); δ _H 5.467 (d, 1.5 Hz, H-1), 3.745 (s, OCH₃). Acetylation of 13 gave diacetate: δ H 5.523 (dd, 1.8, 3.4 Hz, H-2), 5.331 (t, 9.8 Hz, H-4). 14: [α] D-3.0° (c 0.5); δH 5.733 (dd, 1.8, 3.1 Hz, H-2), 5.610 (t, 9.8 Hz, H-4), 3.804 and 3.771 (2d, 14.7 Hz, CH₂Cl). 16: [a] D -3.2° (c 1.4); δ H 5.632 (d, 1.8 Hz, H-1), 5.558 (d, 1.8 Hz, H-1), 5.480 (t, 10.0 Hz, H-4b); δ C 99.2 (¹J_{CH} 175 Hz, C-1), 96.9 (¹J_{CH} 175 Hz, C-1). 17: [α]_D -11.7° (c 1.9); δ _H 5.628 (d, 1.8 Hz, H-1), 5.554 (d, 1.8 Hz, H-1), 4.429 (t, 10 Hz, H-4a), 4.110 (t, 10.0 Hz, H-4b). 18: [a]D -26.4° (c 1.1); δ_H 5.582 (d, 1.8 Hz, H-1 x 2), 5.563 (d, 1.8 Hz, H-1), 5.468 (t, 10.0 Hz, H-4c). 19: δ H 5.556 (d, 1.8 Hz, H-1), 5.506 (bs, H-1), 5.480 (t, 10.0 Hz, H-4c), 5.382 (bs, H-1a). 20: [a]D -18.2° (c 0.6); δ H 5.564 (d, 1.5 Hz, H-1), 5.514 (d, 1.8 Hz, H-1), 5.476 (t, 10.0 Hz, H-4c), 5.326 (d, 1.5 Hz, H-1a), 2.216 (s, SCH3). 21: [a]D -42.6° (c 0.9); δ H 5.587 (bs, H-1 x 2), 5.565 (bs, H-1 x 2), 5.548 (d, 1.8 Hz, H-1), 5.476 (t, 9.8 Hz, H-4f), 5.336 (d, 1.5 Hz, H-1a). 22: 8H 5.330 (s, H-1a), 2.219 (s, SCH₃). 23: 8H 5.244 (d, 1.5 Hz, H-1a), 2.151 (s, SCH₃).
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